

PHY 127 FS2023

Prof. Ben Kilminster

Lecture 4

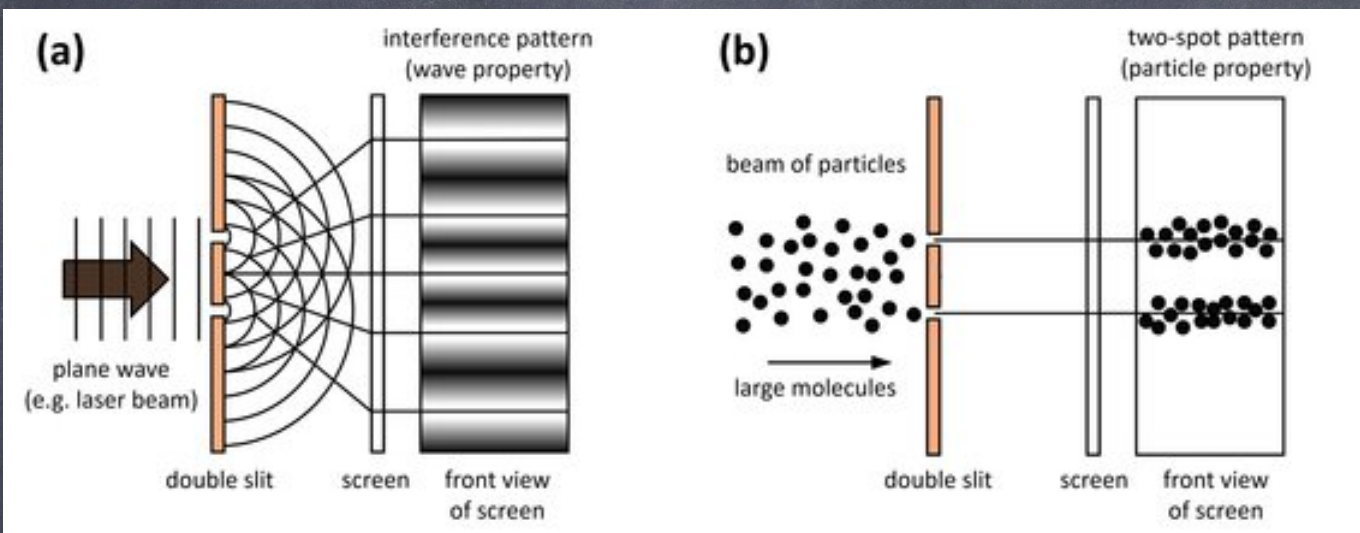
March 17, 2023

Question

Prof.K's new mustache is most like :

	Unanswered	Right	Wrong
Sam Elliot in "The Big Lebowski"	5	15	18
Robert Redford, "The Sundance Kid"	7	6	25
Mandy Patinkin, "The Princess Bride"	5	12	21
Will Ferrell, "Anchorman"	7	10	21
Clark Gable, "Gone with the Wind"	7	6	25
Sacha Baron Cohen, "Borat"	8	12	18





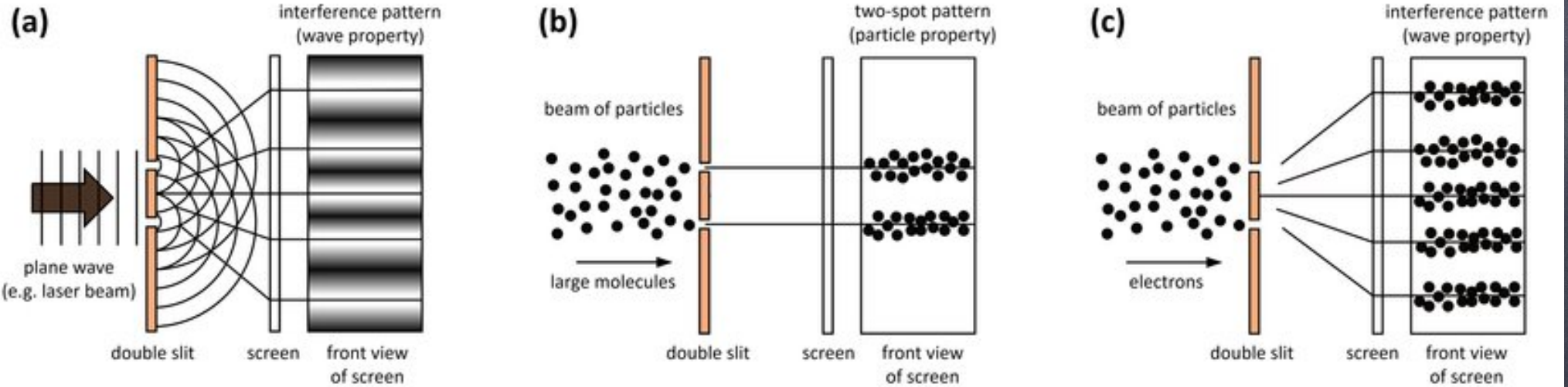
waves



particles

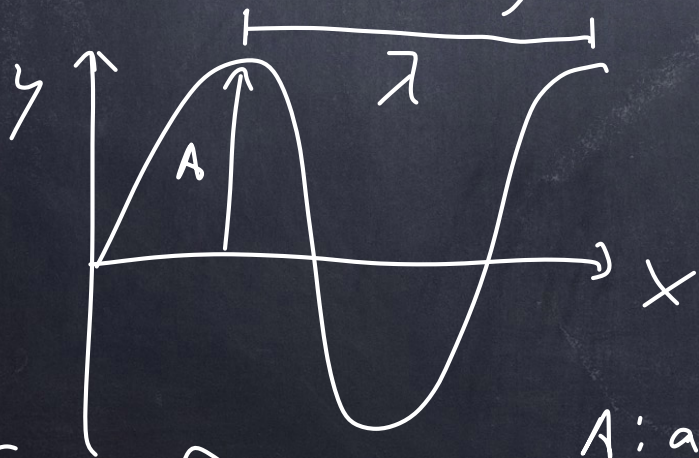


photons



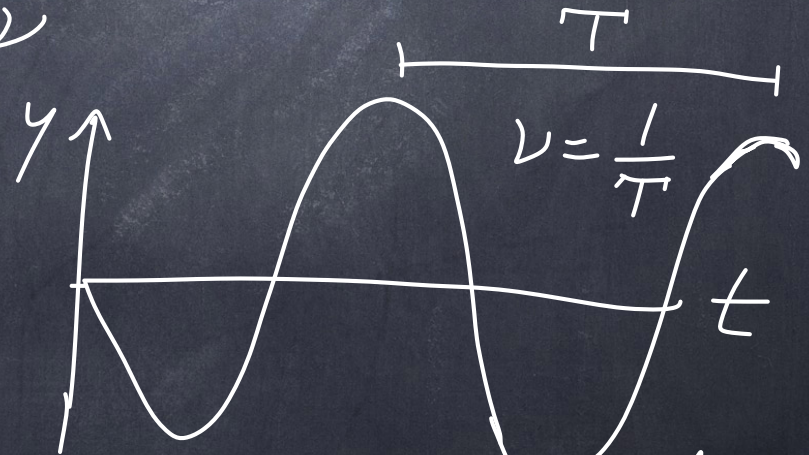
particles are waves and energy is quantized.
 Quantization can be thought of as standing waves. Today we make standing waves. We will see, touch, hear standing waves \rightarrow let's us understand quantized energy levels² of an atom.

To describe a sine wave moving with velocity v , where $v = \lambda \nu$



if we freeze a wave at time $t=0$

A: amplitude



if we look at a point $x=0$

The wave travels a distance λ in a time $T = \frac{1}{\nu}$
the velocity is then $v = \frac{\lambda}{T} = \lambda \nu$

The formula for the wave is called
the wave function:

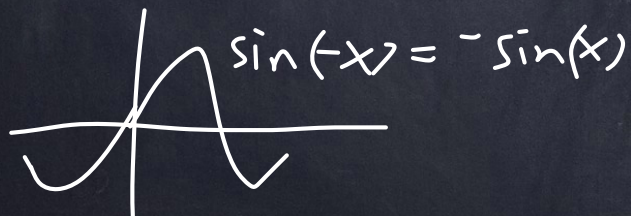
$$y(x, t) = A \sin(kx - \omega t)$$

k : wave number $k = \frac{2\pi}{\lambda}$

ω : angular frequency $\omega = 2\pi\nu = \frac{2\pi}{T}$

If $t=0$, $y(x, t=0) = A \sin kx = A \sin \frac{2\pi x}{\lambda}$

If $x=0$, $y(x=0, t) = A \sin(\omega t) = -A \sin \omega t = -A \sin \frac{2\pi t}{T}$



waves on a string:

depends on tension (F), + the mass density
(force) $\mu = \frac{\text{mass}}{\text{length}}$

Using Newton's Laws, $\Sigma F = ma$, we can derive the wave equation to explain how waves move on a string (see script for physics I, chapter 3)

wave equation

$$\left[\frac{\partial^2 y}{\partial x^2} = \frac{1}{v^2} \frac{\partial^2 y}{\partial t^2} \quad v = \sqrt{\frac{F}{\mu}} \right]$$

The symbol $\frac{\partial}{\partial x}$ means the partial derivative with respect to x .

- y : The height of the wave
- x : the distance along the string
- v : velocity of the wave
- t : the time

$y = y(x, t)$! this means the amplitude (y) depends on x & t .

A solution to the wave equation is
 $y = A \sin(kx - \omega t)$

Standing waves:

A formula for a standing wave is

$$y(x, t) = A \cos \omega t \sin kx \quad] \text{ one solution.}$$

At a time $t=0 \Rightarrow y(x, t=0) = A \sin kx$

string



string is fixed at $x=0$ and $x=L$

The general wave function that is a solution to the wave equation for a standing wave at time $t=0$ is:

$$y(x) = A \sin\left(\frac{2\pi x}{\lambda}\right)$$

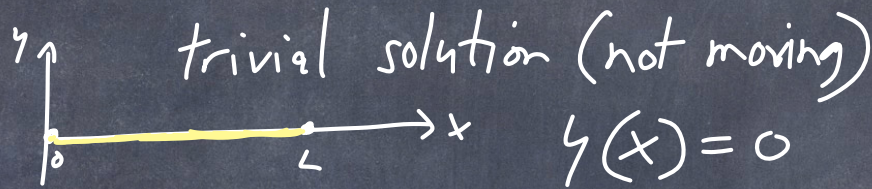
To find the solutions, we substitute different λ as a function of L .

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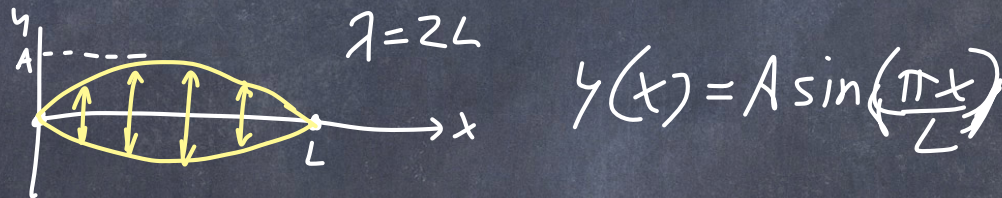
$$y(x) = A \sin\left(\frac{2\pi x}{\lambda}\right)$$

To find the solutions, we substitute different λ as a function of L

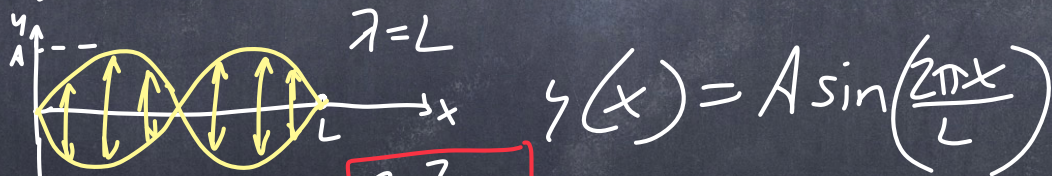
0 bumps



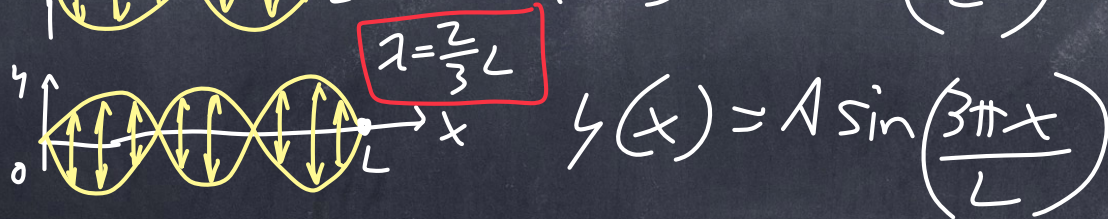
1 bump



2 bumps

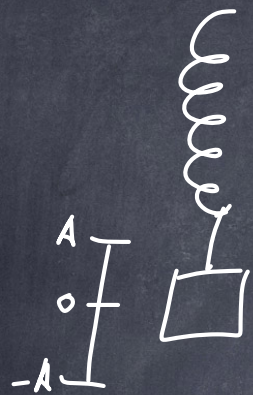


3 bumps



In general, the solutions are $y(x) = A \sin\left(\frac{n\pi x}{L}\right)$ for $n=0, 1, 2, 3, \dots$
 $n = \text{number of bumps}$

Energy transmitted by a wave



A simple harmonic oscillator has an energy of $E = \frac{1}{2}kA^2$

A: amplitude
K: spring constant

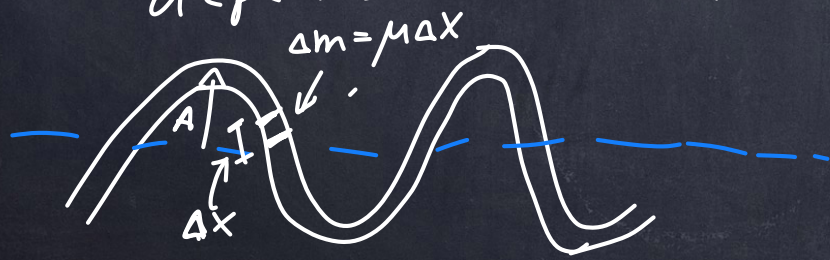
↑
energy

The energy depends on the amplitude squared.

The spring constant is related to the angular frequency by $k = m\omega^2$

m: mass of object
 ω : angular frequency

A string oscillating up and down is also a simple harmonic oscillator. Here the energy depends on $m\omega^2$



$$E = \frac{1}{2}m\omega^2 A^2$$

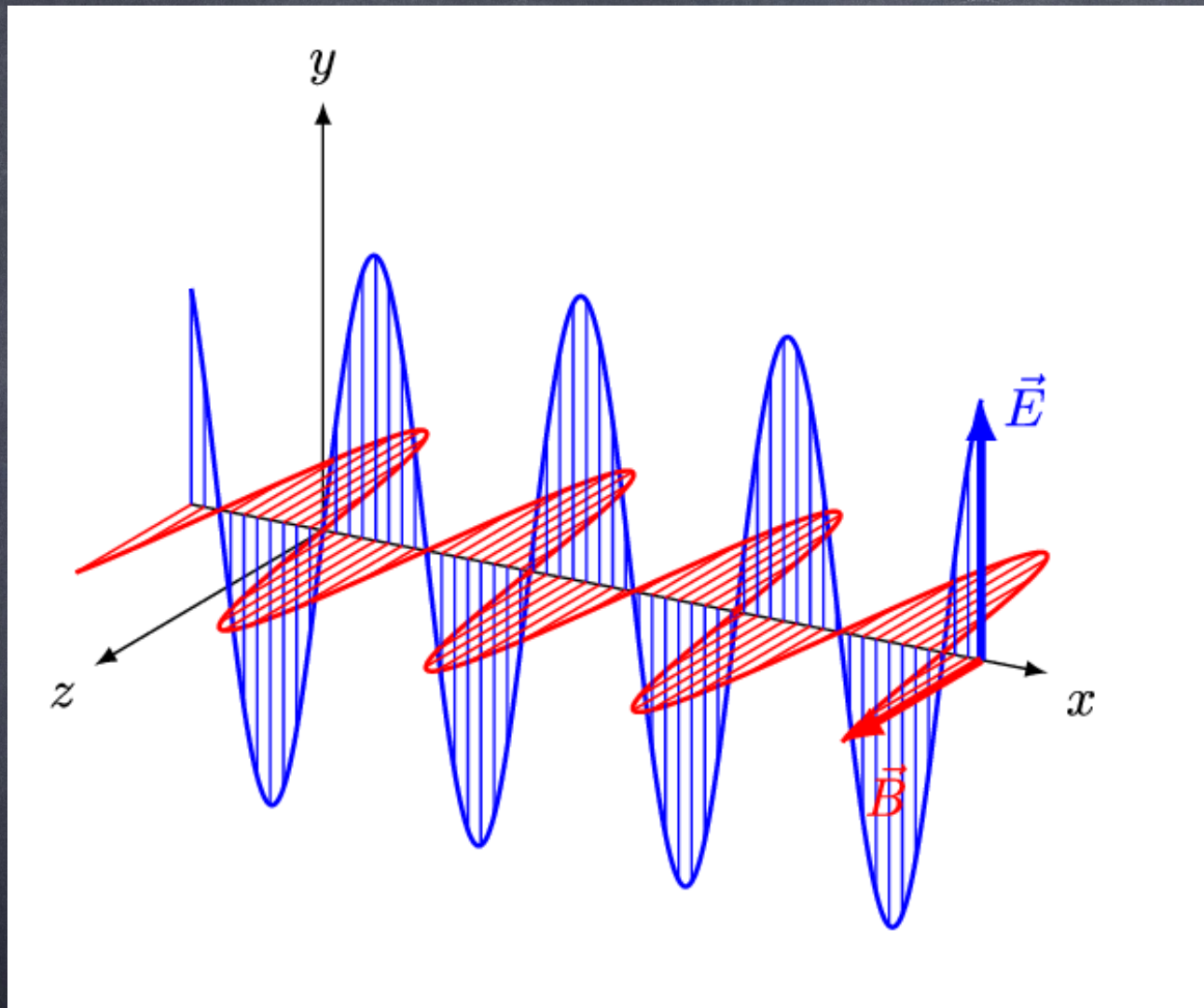
A small part of string carries an energy: $\Delta E = \frac{1}{2}\Delta m\omega^2 A^2 = \frac{1}{2}\mu\omega^2 A^2 \Delta x$

Power transmitted is energy per time:

$$P = \frac{\Delta E}{\Delta t} = \frac{1}{2} \mu \omega^2 A^2 \underbrace{\frac{\Delta x}{\Delta t}}_v = \frac{1}{2} \mu \omega^2 A^2 v$$

Both the energy and power are proportional
to the amplitude squared.

An electromagnetic wave :



\vec{E} : \updownarrow
 \vec{B} : \rightarrow
 \vec{z} : \rightarrow

For an EM wave:



wave functions: $E_y = E_{y_0} \sin(kx - \omega t)$
maximum amplitude

$$B_z = B_{z_0} \sin(kx - \omega t)$$

simple proportionality: $E = cB$

The energy density of the EM wave is

$$\eta_E = \frac{1}{2} \epsilon_0 E^2$$

$$\eta_B = \frac{1}{2} \frac{B^2}{\mu_0} = \frac{1}{2} \epsilon_0 E^2$$

η : energy per unit volume

$$\left\{ \begin{array}{l} c^2 = \frac{1}{\epsilon_0 \mu_0} \\ \downarrow E = cB \end{array} \right.$$

The total energy density is

$$\eta = \eta_E + \eta_B = \epsilon_0 E^2$$

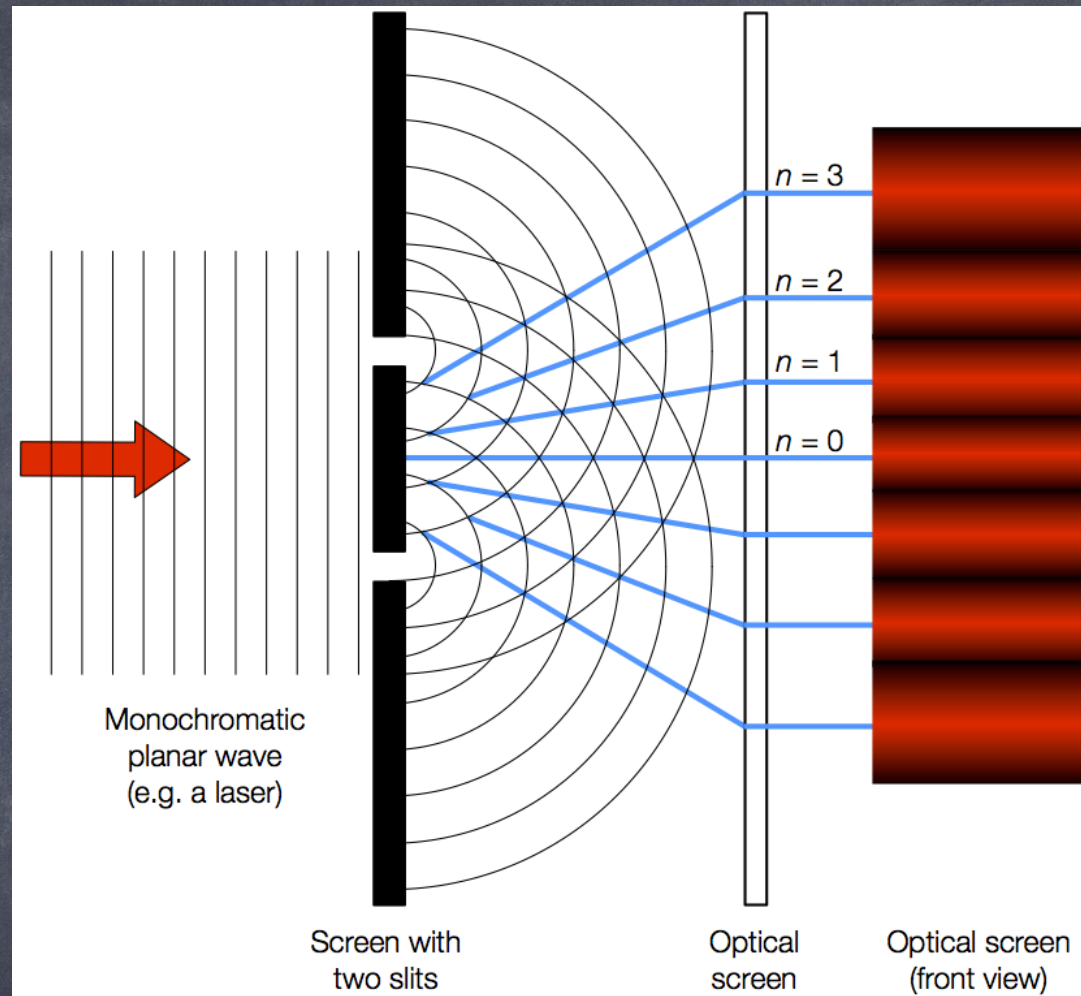
The ~~inter~~ instantaneous intensity I is

$$\frac{\text{power}}{\text{area}} = \text{energy density} * \text{velocity}$$

$$I = \eta c = c \epsilon_0 E^2 = c \epsilon_0 E_0^2 \sin^2(kx - \omega t)$$

The intensity of light is proportional to the square of the electric field.

Remember:



measure of intensity

zero intensity

maximum intensity

Intensity has a special meaning $I \propto E^2$

But this is also a measure of the probability of where we find the photons.

$$P \propto E^2$$

we have $I \propto E(x)^2$ E is the electric field
wave function

$$P(x) \stackrel{\downarrow}{=} \psi^2(x)$$

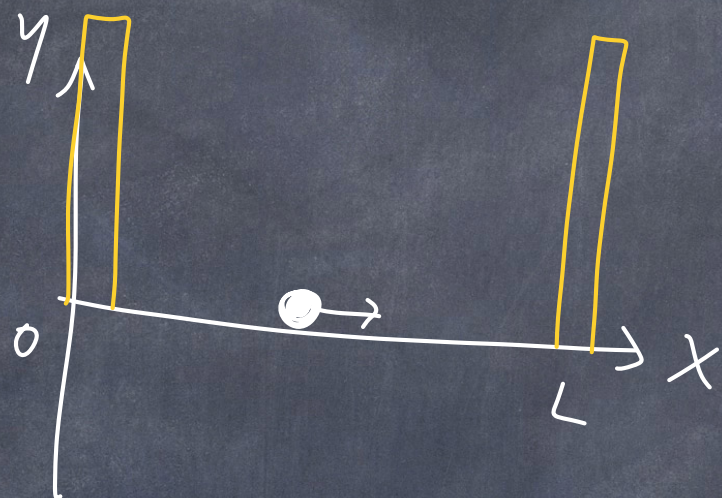
probability

$P(x)$ = probability distribution function

$\psi(x)$ = wave function
(a measure of the amplitude
of the wave)

Consider the classical world (that we know).

Consider a 1-D box with an electron inside.



box has length, L

classically, the electron moves back & forth crashing into walls.

If we know the starting position & velocity, we can predict its location at any time.

If we don't know its starting position, then we only know a probability of where it is.



equally likely to be anywhere in the box.

But it must be in the box.

$$\text{So } \int_0^L P(x) dx = 1$$

$$\int_0^L P dx = 1$$

because is constant
with x .

so we can solve this: $P_x \Big|_0^L = 1$

$$P(L) - P(0) = PL = 1$$

$$P = \frac{1}{L}$$

So actually we want a probability for it to be in a finite space.



$$\begin{aligned} P_{x_1 \rightarrow x_2} &= \int_{x_1}^{x_2} P(x) dx = \int_{x_1}^{x_2} \frac{1}{L} dx \\ &= \left[\frac{x}{L} \right]_{x_1}^{x_2} = \frac{x_2 - x_1}{L} \end{aligned}$$

$$\text{For } x_1 = 0, x_2 = \frac{1}{2}L \Rightarrow \frac{\frac{1}{2}L - 0}{L} = \frac{1}{2}$$

A wave is a particle,
and a particle is a wave.
This is true for any particle.

The wavelength of a particle is $\lambda = \frac{h}{p}$

p : momentum

This is the de Broglie wavelength.

An electron also behaves like a wave.

The energy of a particle : $E = K + U$

$$K = \frac{1}{2}mv^2 = \frac{p^2}{2m}$$

$p = mv$

$$E = \frac{p^2}{2m} + U = \frac{h^2}{\lambda^2 2m} + U$$

kinetic energy

potential energy